

Re: H2O2

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Bernhard Kuemel wrote:

> *Mark Thorson wrote:*

>

>>

>>> *How can hydrogen peroxide exist?*

>>>

>>> *Hydrogen has a +1 and a -1 oxidation states, and oxygen only has a -2.*

>>> *Thus it is impossible to get them to balance with H2O2. So is my*

>>> *oxidation states table in my CRC handbook wrong, or am I missing*

>>> *something?*

>>

>>

>> *While you're at it, how can oxygen gas (O2) exist? :-)*

>

>

> *Or oxygen fluoride OF2*

>

Or even helium trioxide, HeO3

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Dirk

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