

Re: Physical Interpretation of Negative pH.

Source: <http://sci.tech-archive.net/Archive/sci.chem/2005-08/msg00894.html>

- *From:* "Farooq W" <farooq.w@xxxxxxxxx>
 - *Date:* 29 Aug 2005 12:26:26 -0700
-

Borek wrote:

> On Mon, 29 Aug 2005 19:08:59 +0200, Farooq W <farooq.w@xxxxxxxxx> wrote:
>
>>> (iii) A short note in J.Chem.Ed has a procedure for making a pH
>>> -47 to +47 HCl solution which I think is meant to be a joke.
>
>>> I agree with Marvin – that's not even funny.
>
>> Fortunately this is not my view too. The Journal of Chemical Education
>> must be held responsible for spreading such ideas.
>
> Which issue? Perhaps that was intended as April follie?

Pilar, Frank L., Preparation of solutions in the pH range of approximately -47 to +47. J.Chem.Ed., 1972, 49, 776

>
> Best,
> Borek
> --
> <http://www.chembuddy.com> – chemical calculators for labs and education
> BATE – program for pH calculations
> CASC – Concentration and Solution Calculator
> pH lectures – guide to hand pH calculation with examples

• *References:*

- ◆ ***Physical Interpretation of Negative pH.***
 ◇ *From:* Farooq W
- ◆ ***Re: Physical Interpretation of Negative pH.***
 ◇ *From:* Borek
- ◆ ***Re: Physical Interpretation of Negative pH.***
 ◇ *From:* Farooq W
- ◆ ***Re: Physical Interpretation of Negative pH.***

Re: Physical Interpretation of Negative pH.

◇ *From:* Borek

- Prev by Date: ***Re: Physical Interpretation of Negative pH.***
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