

Re: Water – H2O – as a fuel?

Source: <http://sci.tech-archive.net/Archive/sci.energy/2006-02/msg00011.html>

- *From:* Damon Hill <damon1six1@xxxxxxxxxxxx>
 - *Date:* Thu, 02 Feb 2006 05:03:13 -0600
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"George" <K19junk@xxxxxxxxxxxx> wrote in news:1138866919.613598.32150@g14g2000cwa.googlegroups.com:

>> Damon Hill wrote:
>> Your statement seems to be several cookies short of a full package.
>> What's the other 25%? I'd really like to know.
>
> Actually, I read on a website that Water is not exactly 2/3 Hydrogen
> and 1/3 Oxygen, (there are impurities). But if it is more than 50%
> Hydrogen and 25% Oxygen, so much more the 'fuel' for burning, right?
> Thanks for helping me prove my point.

A water molecule is two atoms of hydrogen and one of oxygen.
Any dissolved or suspended materials in a given volume of
water are irrelevant as we were discussing pure water.

The point was that the hydrogen and oxygen in water are
already burnt. That's why water is used to put out fires.
(There are exceptions, as in certain chemical fires.)

And I believe I'm satisfied with this discussion, too.
Having gone from simple chemistry to base comedy, I shall retire
from the debate. Carry on, as I know you will.

--Damon (My litmus test for kookery is positive.)

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