

Re: Mercury-Free Bandwagon Gathers Pace

Source: <http://sci.tech-archive.net/Archive/sci.med.dentistry/2005-08/msg01759.html>

- *From:* clintonz@xxxxxxxxxxx
 - *Date:* 26 Aug 2005 04:48:01 -0700
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>
> As far as the electrolytic response or galvanic effect between inert
> compounds like silver mercury amalgam goes, I don't think the trace
> amounts of mercury are anything in comparison to what we're exposed
> to on a daily basis.

YOur assuming its trace amounts, but published papers already show that is not the case. Your false assumption drives your conclusion.

Foods like fish are loaded with it and there's
> nothing we can do. Chances are that eating a couple of fish dinners
> will expose you to more mercury than all the fillings in one's mouth
> ever will.

Amalgam is already on average the largest source of elemental Hg, not counting corrosion. This was recently published in the NEJM. Your lack of research in the area shows. Furthermore, now we are getting in the realm of solid state a far , cry from circuit theory. It's actually very clear that amalgam is not a stable substance at all, nor is it even well understood. When you can set an upper limit on amalgam corrosion under all conditions or even write an equation for basic processes like Hg vapor emission or spontaneous surface Hg droplet formation on the newer high copper fillings or even characterise the internal phase changes in amalgam over a long period of time let me know.

Until then I would "can" the uniformed ADA tuna analogies.

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- *Follow-Ups:*
 - ◆ **Re: Mercury-Free Bandwagon Gathers Pace**
 - ◇ *From:* billkatz

Re: Mercury-Free Bandwagon Gathers Pace

• **References:**

- ◆ **Mercury-Free Bandwagon Gathers Pace**
◇ From: Keith P Walsh
- ◆ **Re: Mercury-Free Bandwagon Gathers Pace**
◇ From: Robert Morien
- ◆ **Re: Mercury-Free Bandwagon Gathers Pace**
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