

Re: Bohr's Atom still number one

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- *From:* bz <bz+spr@xxxxxxxxxxxxxxxxxxxxxxxx>
 - *Date:* Wed, 2 May 2007 09:24:17 +0000 (UTC)
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"guskz@xxxxxxxx" <guskz@xxxxxxxx> wrote in
news:1178085610.863662.57900@xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx:

On May 1, 5:46 am, bz <bz+...@xxxxxxxxxxxxxxxxxxxxxxxx> wrote:

"g...@xxxxxxxx" <g...@xxxxxxxx> wrote
innews:1177995332.605199.251900@xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx:

(b) the s-orbital is spherical and the Bohr orbit is not

Bohr's orbit isn't circular? then what is it squarical?

Circular is not spherical.

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Bohr's orbitals ALL lie in the same plane.
Bohr's model is disproved by the bonding of molecules.

There is an obedience (non-random and non cloudy/fuzzy) of perfect INTEGER symmetry that is "demanded" in the GEOMETRY of the atom for the Table of Elements to obey the quantum principle and for the electron's necessity to shed or gain light at specific intervals in this geometry.

What 'electron's necessity to shed or gain light at specific intervals'???

It is a no dice game.

Einstein said something similar. You are no Einstein.

Re: Bohr's Atom still number one

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bz

please pardon my infinite ignorance, the set-of-things-I-do-not-know is an infinite set.

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