

Re: Protons & electrons attractions

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- *From:* "newedana" <simplesong1004@xxxxxxxxxxxxx>
 - *Date:* 21 Apr 2005 04:37:45 -0700
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>Anybody can suspect how can orbital electron keep its position around its nucleus without merging to its nucleus. Uncertainty principle was applied for the reason, their merging is prevented. However, I don't believe it is a real science. As one can find in a standard text of QM theory, this uncertainty principle is the first postulation of the theory. Measuring order of the nature by mankind cannot change the scientific truth. In its second postulation, the orbital electron builds an electron cloud around its nucleus, and one can find it with an equal probability. I believe this is rather an ignorant science. QM man believes that orbital electron can float around its nucleus as a wave form like a tiny water droplets of clouds in the sky. However, the standard text says that kinetic energy of orbital electron can be in balance with its potential energy by nuclear attraction, We find also in the Shrodinger's equation that kinetic energy plus potential energy equals constant. However, in order for one to be possible to find the electron forming a three dimensional, spherical cloud with the same probability, the electron has to move at a few thousand times of greater speed than that of Bohr's orbital electron forming a two dimensional ring orbit. If so, it is impossible to balace between its kinetic and potential energy. It is a plain science, it is a basic mistake to apply the probability mathematics in explaining the orbital electron forming a three dimensional spherical cloud. Because it is out of science. newedana

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• *References:*

- ◆ *Protons & electrons attractions*
◇ *From:* Watclod
- ◆ *Re: Protons & electrons attractions*
◇ *From:* newedana

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