

# Present atomic physics is built based on a large number of fraudulent postulations

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Present atomic physics was built based on completely unscientific fraudulent postulations such as:

1. Atomic orbital electron performs a planetary motion around its nucleus, a postulation disregarding the fact that atom does not exist lonely as a single body, and its orbital electron is unable to keep its planetary motion due to interruptive electromagnetic force given out from its neighboring other element particles.

2. Electron is unable to move without inducing magnetism in perpendicular to its moving path. Nevertheless, present atomic model does not account this magnetism to be added to the centrifugal force of orbital electron in balancing to its nuclear Coulomb attraction, since this magnetism acts to straighten linearly its circular orbit. J.J Thomson neglected this magnetism in measuring e/m value, so the present mass of electron was greatly overestimated one.

3. It is a basic postulation one proton can attract one electron in the present atomic model. Nevertheless, a proton can attract 92 electrons if it joins in to a uranium nucleus, so the gross Coulomb attraction between its nucleus and its 92 orbital electrons turns out to be  $92^2 = 8464$  in average, according to the Coulomb law. This self-contradiction of Coulomb law found in current atomic physics is due to its simple imitation of Newton's gravitation law.

4. There does not exist a discontinuous motion of mass particle in the space of nature, because discontinuous motion of a mass particle in scientific sense means that it disappears instantly at a place and then reappears at another place. However, present atomic model has an orbital electron moving by such a discontinuous manner, the quantum jump motion from orbit to orbit.

5. It is a unscientific postulate atomic orbital electron emits photons or lights as it drops down from a higher energy level to a lower one, because potential energy is unable to convert into realistic photons or lights, and such a dropping or a circular motion of orbital electron is not an oscillation possible to generate light with electromagnetic wave phase of counter polarity.

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6. Probability–electron in the quantum atomic model is unable to detect with the same probability in the same electron cloud unlike its postulation. Electron is impossible to cover a three–dimensional x, y, z space by its orbital motion in balancing its kinetic and potential energy at the same moment. Nevertheless, the Schrödinger wave equation was built by ba